



6

(b)  $\text{Cl}_2(\text{g})$  dissolves in cold water and reacts with it.



(i) Identify the oxidation number of chlorine in each of the chlorine-containing species in this reaction.

chlorine-containing species	$\text{Cl}_2$	$\text{HCl}^{\text{+1}}$	$\text{HClO}^{\text{+1} \text{+2}}$
oxidation number of chlorine	0	-1	+1

[2]

(ii) Name the type of reaction occurring.

*solved by TheamKL*

*Disproportionation*

[1]

# June 2013 Chemistry 9701 Paper 22 Question

**Annelies Wilder-Smith**

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