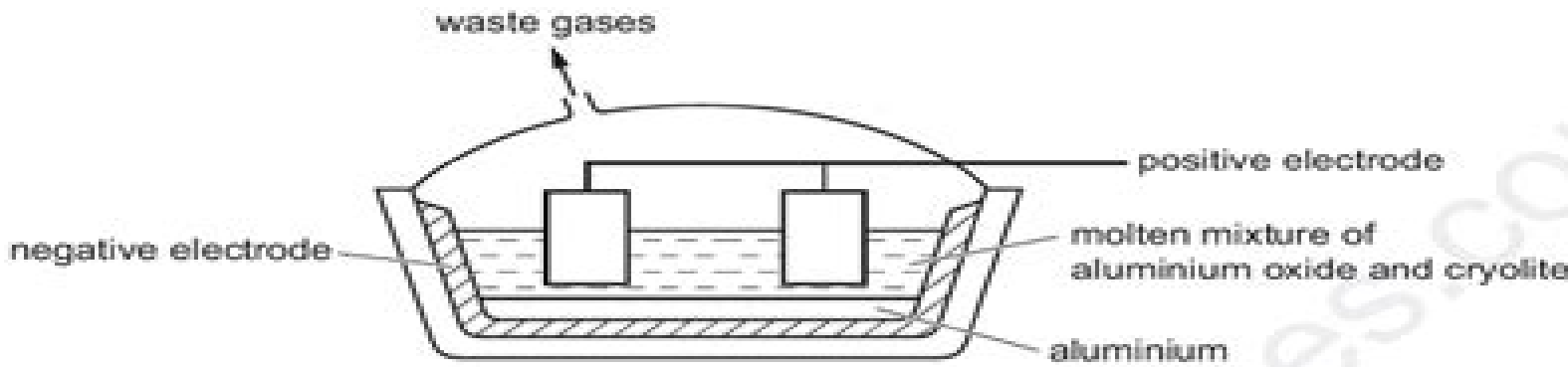


3 (a) Name the ore of aluminium which mainly consists of aluminium oxide.

Bauxite [1]

(b) Aluminium is produced by the electrolysis of aluminium oxide dissolved in molten cryolite.



(i) Give **two** reasons why the electrolysis is done using a molten mixture of aluminium oxide and cryolite instead of molten aluminium oxide only.

- 1 It improves conductivity or It is a better conductor
 - 2 It has a lower operating temperature.
- [2]

(ii) Write ionic half-equations for the reactions occurring at the electrodes.

positive electrode $2\text{O}^{2-} \rightarrow \text{O}_2 + 4\text{e}^-$

negative electrode $\text{Al}^{3+} + 3\text{e}^- \rightarrow \text{Al}$

[2]

(iii) The anodes are made of carbon and have to be replaced regularly.

Explain why the carbon anodes have to be replaced regularly.

Anodes(or carbon) react with oxygen to form carbondioxide.Hence they need to be replaced.

[2]

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Table of Contents Igcse Question Papers Chemistry

1. Understanding the eBook Igcse Question Papers Chemistry
 - The Rise of Digital Reading Igcse Question Papers Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Igcse Question Papers Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Igcse Question Papers Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Igcse Question Papers Chemistry
 - Personalized Recommendations
 - Igcse Question Papers Chemistry User Reviews and Ratings
 - Igcse Question Papers Chemistry and Bestseller Lists
5. Accessing Igcse Question Papers Chemistry Free and Paid eBooks
 - Igcse Question Papers Chemistry Public Domain eBooks
 - Igcse Question Papers Chemistry eBook Subscription Services
 - Igcse Question Papers Chemistry Budget-Friendly Options

6. Navigating Igcse Question Papers Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Igcse Question Papers Chemistry Compatibility with Devices
 - Igcse Question Papers Chemistry Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Igcse Question Papers Chemistry
 - Highlighting and Note-Taking Igcse Question Papers Chemistry
 - Interactive Elements Igcse Question Papers Chemistry
8. Staying Engaged with Igcse Question Papers Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Igcse Question Papers Chemistry
9. Balancing eBooks and Physical Books Igcse Question Papers Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Igcse Question Papers Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Igcse Question Papers Chemistry
 - Setting Reading Goals Igcse Question Papers Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Igcse Question Papers Chemistry
 - Fact-Checking eBook Content of Igcse Question Papers Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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web testout has no affiliation with these companies and the products and services advertised herein are not endorsed by any of them table of contents2 1 1 network pro introduction4 1 2 use the simulator5 networking basics6 2 1 networking overview6 2 2 osi model and data encapsulation9 2 3 data encapsulation11 2 4 network protocols13

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