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Pearson Edexcel International GCSE

Time 2 hours 30 minutes

Paper
reference

4MB1/02R

Mathematics B PAPER 2R



You must have: Ruler graduated in centimetres and millimetres, protractor, pair of compasses, pen, HB pencil, eraser, calculator. Tracing paper may be used.

Total Marks

Instructions

- Use **black** ink or ball-point pen.
- **Fill in the boxes** at the top of this page with your name, centre number and candidate number.
- Answer **all** questions.
- Answer the questions in the spaces provided
– *there may be more space than you need.*
- **Calculators may be used.**

Information

- The total mark for this paper is 100.
- The marks for **each** question are shown in brackets
– *use this as a guide as to how much time to spend on each question.*

Advice

- Read each question carefully before you start to answer it.
- Check your answers if you have time at the end.
- Without sufficient working, correct answers may be awarded no marks.

Turn over ➤

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Igcse Edexcel Maths B Jan 2015 Question Paper 2r

**Jan Dangerfield, Rose Jewell, Sue
Pope, Nick Geere**



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 - Chapter 5: Conclusion
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 6. In chapter 5, this book will draw a conclusion about Igcse Edexcel Maths B Jan 2015 Question Paper 2r. This chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Igcse Edexcel Maths B Jan 2015 Question Paper 2r.

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web limiting reagents and percentage yield worksheet 1 consider the reaction $\text{I}_2\text{O}_5 + \text{CO} \rightarrow \text{CO}_2 + \text{I}_2$ a 80.0 grams of iodine v oxide i₂o₅ reacts with 28.0 grams of carbon monoxide co determine the mass of iodine i₂ which could be produced 80 g i₂o₅ 1 mol i₂o₅ 1 mol i₂ xs 1 333.8 g i₂o₅ 1 mol i₂o₅ 28 g co 1 mol co

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web jun 2 2020 the percent yield of a reaction is the ratio of the actual yield to the theoretical yield multiplied by 100 to give a percentage text percent yield text actual yield g over text theoretical yield g times 100 label 3 7 3