

Matter is composed of organic and inorganic compounds

- **Hydrocarbons** = contain only carbon and hydrogen
 - The simplest hydrocarbon is methane (natural gas)
 - Hydrocarbons can be a gas, liquid, or solid
- Fossil fuels consist of hydrocarbons
 - Some can be harmful to wildlife

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Brendan G. Carr



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Fredeen, Pearson, 2003-07 *Leg Ol Sci Chem* Pearson Education, Limited, 2007-10-31 *The Pearson Guide to Inorganic Chemistry for the IIT JEE 2012*, *Commonly Asked Questions in Thermodynamics* Marc J. Assael, Geoffrey C. Maitland, Thomas Maskow, Urs von Stockar, William A. Wakeham, Stefan Will, 2022-08-05 CRC Press is pleased to introduce the new edition of Commonly Asked Questions in Thermodynamics an indispensable resource for those in modern science and engineering disciplines from molecular science engineering and biotechnology to astrophysics Fully updated throughout this edition features two new chapters focused on energy utilization and biological systems This edition begins by setting out the fundamentals of thermodynamics including its basic laws and overarching principles It provides explanations of those principles in an organized manner using questions that arise frequently from undergraduates in the classroom as the stimulus These early chapters explore the language of thermodynamics the first and second laws statistical mechanical theory measurement of thermodynamic quantities and their relationships phase behavior in single and multicomponent systems electrochemistry and chemical and biochemical reaction equilibria The later chapters explore applications of these fundamentals to a diverse set of subjects including power generation with and without fossil fuels for transport industrial and domestic use heating decarbonization technologies energy storage refrigeration environmental pollution and biotechnology

Data sources for the properties needed to complete thermodynamic evaluations of many processes are included. The text is designed for readers to dip into to find an answer to a specific question where thermodynamics can provide some if not all of the answers whether in the context of an undergraduate course or not. Thus its readership extends beyond conventional technical undergraduates to practicing engineers and also to the interested lay person who seeks to understand the discourse that surrounds the choice of particular technological solutions to current and future energy and material production problems. *The IIT Foundation Series - Chemistry Class 9, 2/e*, *The Pearson CSAT Manual 2012* Edgar Thorpe, 2012

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