

## Chapter 4 Atomic Structure

**Section 4.2 The Structure of an Atom****(pages 108–112)**

*This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.*

**Reading Strategy (page 108)**

**Monitoring Your Understanding** Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

| What I Know About Atoms                                                                                                           | What I Would Like to Learn                                                                                     | What I Have Learned |
|-----------------------------------------------------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------------------|---------------------|
| Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles. | Based on the title of the section, students may say that they want to learn more about the structure of atoms. |                     |

**Properties of Subatomic Particles (pages 108–109)**

- What are three subatomic particles?  
a. Protons      b. Electrons      c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.  
a. nucleus      **(b)** proton  
c. neutron      d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.  
\_\_\_\_\_  
\_\_\_\_\_

**Comparing Subatomic Particles (pages 109–110)**

- Circle the letters of properties that vary among subatomic particles.  
a. color      **(b)** mass  
**(c)** charge      **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.  
**(a)** mass of 1 neutron = mass of 1 proton  
b. mass of 2000 neutrons = mass of 1 proton  
c. mass of 1 electron = mass of 1 proton  
d. mass of 1 neutron = mass of 1 electron

# Hewitt Chapter 4 Worksheet Answers

**Ying Liu**



## **Hewitt Chapter 4 Worksheet Answers:**

## Reviewing **Hewitt Chapter 4 Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Hewitt Chapter 4 Worksheet Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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genesis 27 1 28 7 years later jacob and esau met and were reconciled genesis 33 both jacob and esau were fathers of nations  
god changed jacob s name to israel genesis 32 28 and he became the father of the 12 tribes of israel

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web the lord wanted jacob to have the birthright blessing because esau was not worthy of it jacob asked esau to trade him the birthright for some food esau agreed and traded his birthright to jacob genesis 25 23 29 34 hebrews 11 20 image  
rebekah and isaac rebekah and isaac wanted what was best for their children

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