

FunBased Learning

Go to: Class Website → Used in Class → "Fun Based Learning"

Directions:

1. ...Go to Classic ChemBalancer
[when done with this activity, go to Review & Brain Boggle]
2. Click 'Directions'. Read and understand the directions.
3. Click 'OK', then on 'Start Game'
4. Try entering some numbers in the text boxes in front of each molecule.
What happens?
5. If you forget the directions, click on the 'How to Play the Game' link.
Click 'OK' when you finish reading them to return to the game.
6. When you think you have typed the right numbers in all the boxes,
click the 'Balanced' button.
7. If you didn't get it right, try again.
8. If you did get it right, then fill in the correct answers on this worksheet for #1.
9. Repeat steps 7-9 for the other 10 questions.
10. Now do the two problems on the back of this worksheet.

You can draw the molecules just like the program did to figure out the answer.
[11. When you're done, use the line in front of each problem to classify it:
S = synthesis, DR = double replacement, SR = single replacement,
A/R = add/base, D = decomposition, C = combustion]

Classic ChemBalancer Questions

Fill in the blanks below as you go through the game.
This is so I have a record that you did your assignment.

- "_____" 1. ____ Fe + ____ S → ____ FeS
- "_____" 2. ____ H₂ + ____ Cl₂ → ____ HCl
- "_____" 3. ____ Mg + ____ O₂ → ____ MgO
- "_____" 4. ____ O₂ + ____ H₂ → ____ H₂O
- "_____" 5. ____ H₂O → ____ H₂ + ____ O₂
- "_____" 6. ____ Ca + ____ H₂O → ____ Ca(OH)₂ + ____ H₂

- "_____" 7. ____ CH₄ + ____ O₂ → ____ CO₂ + ____ H₂O
- "_____" 8. ____ Na₂O₂ + ____ H₂SO₄ → ____ Na₂SO₄ + ____ H₂O₂
- "_____" 9. ____ N₂ + ____ H₂ → ____ NH₃
- "_____" 10. ____ Al + ____ O₂ → ____ Al₂O₃
- "_____" 11. ____ KMnO₄ → ____ K₂O + ____ MnO + ____ O₂

Draw the molecules just like the program did to figure out the answer to #1, and #11.

- "_____" 12. ____ Na + ____ H₂O → ____ NaOH + ____ H₂
Fact for #12: Sodium metal, Na, is stored in kerosene so it won't react with water vapour. When added to water it reacts quickly to make hydrogen gas.

- "_____" 13. ____ H₂SO₄ + ____ NaOH → ____ Na₂SO₄ + ____ H₂O
Fact for #13: This is an example of an acid base reaction. Acid + base → Salt + Water

Review ChemBalancer Questions

- "_____" 1. ____ S + ____ O₂ → ____ SO₂
- "_____" 2. ____ Na + ____ O₂ → ____ Na₂O₂
- "_____" 3. ____ H₂ + ____ O₂ → ____ H₂O
- "_____" 4. ____ Ag₂O → ____ Ag + ____ O₂
- "_____" 5. ____ Ba(OH)₂ + ____ H₃PO₄ → ____ BaHPO₄ + ____ H₂O
- "_____" 6. ____ NaOH + ____ H₃PO₄ → ____ Na₃HPO₄ + ____ H₂O

Fun Based Learning Review Chembalancer Answers

Sebastian Brünink



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