

Name: _____

WRITING NEUTRAL FORMULAS FOR IONIC COMPOUNDS

Directions: Use a periodic table. Remember that A groups have the following ionic charges: Group 1 (+1), Group 2 (+2), Group 3 (+3), Group 5 (-3), Group 6 (-2), and Group 7 (-1). Transition metals can have variable positive charge, which is given to you in the compound name (ex. Iron (III) has a +3 charge whereas Copper (II) has a +2 charge).

#	Compound Name	Cation	Anion	Neutral Ionic Formula
1	Sodium chloride	Na ⁺	Cl ⁻	NaCl
2	Calcium fluoride	Ca ²⁺	F ⁻	CaF ₂
3	Aluminum chloride			
4	Aluminum oxide			
5	Beryllium fluoride			
6	Sodium oxide			
7	Copper (II) sulfide			
8	Aluminum fluoride			
9	Manganese (III) oxide			
10	Magnesium nitride			
11	Sodium sulfide			
12	Iron (III) oxide			

Formulae Of Ionic Compounds Answers

Hussin A.Rothana

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Formulae Of Ionic Compounds Answers:

Decoding **Formulae Of Ionic Compounds Answers**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Formulae Of Ionic Compounds Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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