

Solution
CHAPTER TEST
Class 11 - Chemistry

1. (a) Propene

Explanation: Elimination reaction takes place, a molecule of HBr gets eliminated to form propene.

2. (c) $\text{HI} > \text{HBr} > \text{HCl}$

Explanation: $\text{HI} > \text{HBr} > \text{HCl}$

3. (a) 1,3-dimethylbenzene



4. (d) Geometric isomers

Explanation: Isomers which have the same structural formulae but differ in the relative spatial arrangements of atoms or groups around the double bond are called geometric isomers. Thus geometrical isomers are a type of stereoisomers.

5. (d) HgSO_4

Explanation: Alkynes are not readily hydrated due to their low reactivity towards electrophilic addition reactions. Further dilute H_2SO_4 has no effect on carbon carbon triple bond. In the presence of HgSO_4 as catalyst, hydration occurs readily.

6. (a) A mixture of low boiling hydrocarbons.

Explanation: Pyrolysis is the decomposition of a compound by the application of heat. It involves the breaking of C-C and C-H bonds resulting in the formation of lower boiling alkanes.

7. (a) 3

Explanation: The position isomers of C_5H_8 are n-pentyne, 3-methyl butyne, pent-2-yne.

8. (a) C_8H_{18}

Explanation: Alkenes are unsaturated hydrocarbons with at least one double bond between the carbon atoms. Their general formula is C_nH_{2n} .

9. (b) Sodium acetate

Explanation:

This is an example of Kolbe's electrolysis method. The reaction is:



The step-wise redox reactions occurring in the electrolytic cell are depicted as under



At Anode: (Oxidation)



At Cathode: (Reduction)



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1. Understanding the eBook Hydrocarbons Answer Key
 - The Rise of Digital Reading Hydrocarbons Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbons Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hydrocarbons Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbons Answer Key
 - Personalized Recommendations
 - Hydrocarbons Answer Key User Reviews and Ratings
 - Hydrocarbons Answer Key and Bestseller Lists

5. Accessing Hydrocarbons Answer Key Free and Paid eBooks
 - Hydrocarbons Answer Key Public Domain eBooks
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 - Hydrocarbons Answer Key Enhanced eBook Features
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 - Highlighting and Note-Taking Hydrocarbons Answer Key
 - Interactive Elements Hydrocarbons Answer Key
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hydrocarbons Answer Key
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 - Creating a Diverse Reading Collection Hydrocarbons Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Hydrocarbons Answer Key
 - Setting Reading Goals Hydrocarbons Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hydrocarbons Answer Key
 - Fact-Checking eBook Content of Hydrocarbons Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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