

## Unit 18: Hydrocarbon and its Compounds

### Things to be Studied in this Unit

- Introduction to hydrocarbon
- Saturated and unsaturated hydrocarbon
- Alkane, alkene and alkyne
- Homologous series
- Alkyl radical
- Functional group
- IUPAC system
- Common system to nomenclature of hydrocarbons
- Some hydrocarbon and its compounds
  - Methane
  - Propane
  - Alcohol
  - Ethane
  - Butane
  - Glycerol

### Introduction

Hydrocarbons are organic compounds composed of hydrogen (H) and carbon (C) atoms. They are of two main types, viz. saturated and Unsaturated. The hydrocarbons in which carbon atoms are combined with a single bond are called saturated hydrocarbons. The hydrocarbons which contain a double bond or triple bond between carbon atoms are called unsaturated hydrocarbons. Hydrocarbons are the basis for many organic compounds and are found in various forms in fossil fuels like mineral oil and natural gas. They serve as essential building blocks for a wide range of chemical products and fuels, playing a vital role in industry and energy production.

### Things to Know

-  **Hydrocarbon:** Hydrocarbons are the compounds formed by the combination of hydrogen and carbon. Examples: Methane ( $\text{CH}_4$ ), Propane ( $\text{C}_3\text{H}_8$ ) etc.
-  **Unsaturated hydrocarbon:** The hydrocarbon which contains a double bond or triple bond between carbon atoms is called unsaturated hydrocarbon. The general formula of unsaturated hydrocarbons is  $\text{C}_n\text{H}_{2n}$  (alkene) and  $\text{C}_n\text{H}_{2n-2}$  (alkyne).
-  **Saturated hydrocarbon:** The hydrocarbon in which carbon atoms are combined with a single bond is called saturated hydrocarbon. The general formula of saturated hydrocarbon is  $\text{C}_n\text{H}_{2n+2}$ .
-  **IUPAC:** IUPAC stands for International Union of Pure and Applied Chemistry.
-  **Alkane:** An alkane is a saturated hydrocarbon in which carbon atoms are connected by only single covalent bonds. The general formula of alkanes is  $\text{C}_n\text{H}_{2n+2}$  where 'n' is the number of carbon atoms present in one molecule of alkane. Example: Propane, Pentane, etc.
-  **Alkene:** An alkene is a saturated hydrocarbon in which carbon atoms are connected by a double bond. So an alkene contains  $-\overset{\textstyle |}{\text{C}}=\overset{\textstyle |}{\text{C}}-$  group. The general formula of an alkene is  $\text{C}_n\text{H}_{2n}$  where 'n' is the number of carbon atoms in its one molecule. Example: Propene, Pentene, etc.

# Hydrocarbon Compounds Guide Answers

**R Sanford**



## **Hydrocarbon Compounds Guide Answers:**

## Reviewing **Hydrocarbon Compounds Guide Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Hydrocarbon Compounds Guide Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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