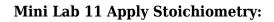
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Substance	Mass (g)	Volume (mL)
Aluminum	13.3 g	
Water	22.8 g	22.8 mL

$$gH_2O = \frac{22.8 \text{ mL } H_2O}{1.00 \text{ g} H_2O} = 22.8 \text{ g} H_2O$$

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