

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The particles of the gases are different sizes.)
- At standard atmospheric conditions, all gases have the same average kinetic energy.
- At standard atmospheric conditions, all gases have the same average speed of particles.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of particles in three phases: solid, liquid, and gas.



Figure 1: Representation of particles in three phases: solid, liquid, and gas.

At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The particles of the gases are different sizes.)

Key Concepts

1. All gases have the same average kinetic energy at standard atmospheric conditions, but are made of different molecules and are made up of different numbers of particles. (The particles of the gases are different sizes.)
2. At standard atmospheric conditions, all gases have the same average speed of particles.

Table 1: Representation of particles in three phases: solid, liquid, and gas.

	SOLID	LIQUID	GAS
How the particles are arranged:	Particles are packed closely together in a regular, repeating pattern. The particles are in contact with each other.	Particles are packed closely together in a disordered, repeating pattern. The particles are in contact with each other.	Particles are widely spaced and move rapidly in all directions. The particles are not in contact with each other.
How the particles move:	Particles vibrate in place. They have a fixed position and can only move back and forth.	Particles move more rapidly than in the solid state. They can move in any direction and can change position.	Particles move very rapidly in all directions. They have a high average speed and can move in any direction.

Lw Pogil Kinetic Molecular Theory Key

R Barnett



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Enjoying the Beat of Phrase: An Emotional Symphony within **Lw Pogil Kinetic Molecular Theory Key**

In a global eaten by screens and the ceaseless chatter of instant conversation, the melodic elegance and emotional symphony created by the written term often diminish in to the back ground, eclipsed by the relentless noise and interruptions that permeate our lives. However, situated within the pages of **Lw Pogil Kinetic Molecular Theory Key** a wonderful fictional value filled with fresh feelings, lies an immersive symphony waiting to be embraced. Crafted by a masterful musician of language, this interesting masterpiece conducts viewers on a mental trip, well unraveling the concealed tunes and profound influence resonating within each cautiously constructed phrase. Within the depths of this moving evaluation, we can explore the book is central harmonies, analyze their enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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