

Who Did It?

Correctly match the person to their accomplishment. Each name may be used more than once.

Bohr Chadwick Dalton Democritus Rutherford Thompson

10. Discovered the nucleus _____
11. Responsible for naming the atom _____
12. Created an atomic theory _____
13. Discovered the proton _____
14. Discovered atoms were mostly empty space _____
15. Discovered the electron _____
16. Worked with Rutherford to discover the neutron _____
17. Concluded that electrons move in orbits _____
18. Stated that atoms could combine to form elements _____
19. First known person to believe that matter was made up of tiny indivisible pieces _____
20. Proposed (but didn't prove) the existence of a positively charged particle _____

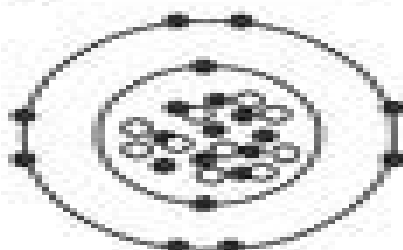
Dalton's Atomic Theory

- Atoms can't be broken into smaller pieces.
- In any element, all the atoms are exactly alike.
- Atoms of different elements are different.
- Atoms of the same element can combine to form compounds.
- Atoms of each element have a unique mass.
- The masses of the atoms in a compound are always in a constant ratio.

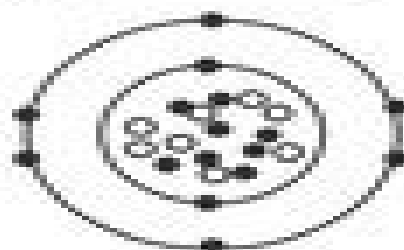
Identify It

Atoms can be identified by counting the number of protons. Using the The Periodic Table of Elements to help you, identify each of the atoms below.

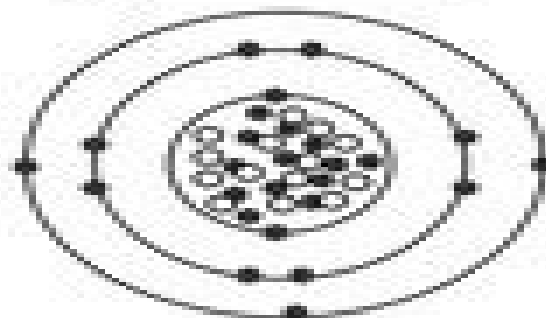
21. _____



22. _____



23. _____



Dalton's Atomic Theory: True or False?

24. All the atoms of any one type of element (like silver) are exactly alike. _____
25. Atoms can't be broken into smaller pieces and still retain the properties of that element. _____

Introduction To Atoms Answer Key

Hussin A.Rothana

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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