

Introduction

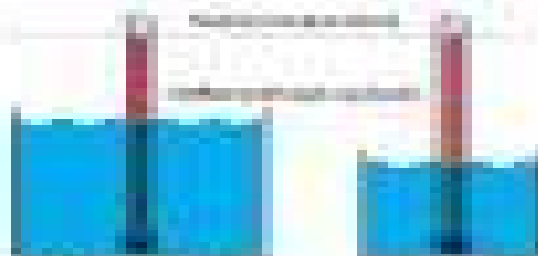
Many natural and man-made systems behave naturally with regard to thermodynamics (in fact, we usually do). Thermodynamics of this can be defined as: _____

Thermodynamics is the study of the _____, the energy interactions that occur within a system boundary and the surroundings (all included thermodynamically). Therefore:

Thermodynamics and its definition

- It forms a primary calculation parameter in thermodynamics (although, not really, it's about).
- Heat: Q
- Temperature: T
- Entropy: S

Heat	Temperature
• Heat is a form of energy transfer across a system boundary to a heat source. • e.g. you feel heat on a hot object (heat is transferred from the _____ to the _____) (heat is transferred from the _____ to the _____).	• Temperature is the average energy that "particles" have in a substance.



Temperature is the average energy that "particles" have in a substance. Temperature: _____
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Definition

- It is a science of heat.
- It is the _____
- It is the study of energy and its conversion to work and heat in a thermodynamic system.



Intro To Thermochemistry Activities

VM Jensen



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