

Internship Application Essay Example

[Your Name]
[Address]
[Contact Details]

Date:

Mr./Ms. [Name]
Company Name:
Address of the company:

Dear Mr./Ms.,

I am writing to apply for this internship program. My solid obligation to my works has been shown in my various volunteer encounters locally and at college. All through my certification, I chipped in week by week at nearby elementary schools where I lectured Class 6 and 7 subjects. Seeing the students light up with fervor while taking an interest in a scientific exploration or the understudy who has battled with a numerical idea to finally get what to do has been limitlessly compensating for me. Past the prizes, these encounters have shown me how to tailor my way to deal with the students' necessities and viably impart ideas to a gathering. abilities I know would work well for me in my future positions.

I have a real premium in learning however much as could be expected about a vocation in business and finance and would see any errand put before me as a positive learning experience. Through my examination as a math understudy, I have demonstrated my capacity to keep up with high interest, learn convoluted things and not surrender without any problem.

Capable in factual, business information investigation, and dominates in cooperative group based conditions. I trust it is an ideal method to use my abilities to take care of business and monetary issues.

My essential objective in starting a profession in business and money is to obtain however much experience as could reasonably be expected. Not just I desire to develop my insight in current business and monetary universes, yet I would likewise want to secure a portion of the functional abilities that I should be better prepared for in this field.

I accept that working in business and money, I would have the advantage to see the value in that everything on the planet by one way or another influences my work. I would trust that this entry-level position would give me a chance to assemble a functioning relationship with pioneers in the field from whom I could acquire important information. I'm sure that any experience from this temporary job would end up being imperative to me in achieving my future objectives. Thank you for your consideration.

Sincerely,
[Your name]

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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values to determine the heat of reaction for the reaction C_2H_4

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web thus for reaction 9 2 in which 1 mole of H_2O and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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