

SNC2D1: Grade 10 Academic Science

Chemistry Test date: _____

Study tips:

- € Read through your notes
- € Make point form notes to summarize the topics
- € Complete the review sheet
- € For extra practice . . .

Do the following **textbook review questions**:

p. 151 – Q. 1, 2, 3, 5, 6, 7

p. 158 – Q. 1, 2, 3, 5, 6, 7

p.168 – Q. 3, 7, 8

Chap. Review p.174-175 – Q.9, 10, 11, 12, 13, 16, 20

p. 189 – Q. 1, 4, 5, 7, 8

p. 198 – Q. 1, 4, 6, 7

p.206 – Q. 1, 4, 7

Chap. Review p.214-215 – Q.1-7, 8, 10, 11, 12, 13, 15, 17

p. 228 – Q. 1, 3, 4, 5, 6, 7, 8

p. 235 – Q. 1, 2, 3, 4, 6, 7, 8

p.246 – Q. 1, 2, 4

Chap. Review p.252-253 – Q.1-7, 8, 9, 10, 11, 12, 13, 16, 18, 19, 20, 21, 22, 26, 28

Chemistry Topics Review

1. Lesson 1A: Lab Equipment
2. Lesson 1B: Safety Symbols & Lab Safety Rules
3. Lesson 2A: Classification of Matter
4. Lesson 2B: Physical & Chemical Change
5. Lesson 3: The Periodic Table, Atomic Structure & Bohr-Rutherford Diagrams
6. Lesson 4: Atoms & Ions / Isoelectric
7. Lesson 5: Lewis Dot Diagrams & Combining Capacity
8. Lesson 6A & 6B: Binary - Ionic Compounds
9. Lesson 7: Molecular Compounds
10. Lesson 8: Ternary - Polyatomic Compounds
11. Lesson 9: Binary Acids & Oxyacids
12. Lesson 10A & 10B: Chemical Reactions & Balancing Equations
13. Lesson 11: Types of Chemical Reactions
14. Lesson 12A & 12B & 12C: Acids & Bases, The pH scale, Neutralization Reaction

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Richard Bailey



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