

Names: _____

Class: _____

Government Study Guide

Write the letter of the correct answer on the line. (Bolded words in Booklet)

- | | | |
|-----------------------|---------------|-------------------|
| A. Law | D. Citizen | G. Constitution |
| B. Government Service | E. Veto | H. Bill of Rights |
| C. Tax | F. Amendments | I. Preamble |

1. I The introduction to the Constitution.
2. D A person who lives in a community.
3. H Lists the ten basic rights all citizens have.
4. B Work that is done to help everyone in the community.
5. E Refusing to sign a law.
6. G This contains the most important laws that everyone must follow.
7. A A rule that everyone in a community, state, or country is expected to follow.
8. F These state the changes to the Constitution.
9. C Money that people and businesses pay to a government.
10. How does the government get money to pay for government services?
☒ A. By collecting taxes from people
B. By holding bake sales
C. By printing their own money
D. By working at a job to earn money
11. Why do communities have governments?
☒ A. to protect their citizens
B. to have elections every four years
C. to give the police and judges a job
D. to have power over the people
12. Who is the leader in the executive branch of a national government? (Executive Branch)
☒ A. President
B. Mayor
C. Governor
13. What is the process of adding an amendment to the Constitution? (Constitution)

**3/4 of the states must approve of
the amendment**

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Judd E. Hollander



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web jun 2 2020 the reactant that is consumed first and limits the amount of product s that can be obtained is the limiting reactant to identify the limiting reactant calculate the number of moles of each reactant present and compare this ratio to the mole ratio of the reactants in the balanced chemical equation

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web jun 30 2023 the limiting reagent is the one that is totally consumed it limits the reaction from continuing because there is none left to react with the in excess reactant there are two ways to determine the limiting reagent

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web a which chemical is the limiting reactant b how many grams of zns will be formed c how many grams of the excess reactant will remain after the reaction is over 3 which element is in excess when 3 00 grams of mg is ignited in 2 20 grams of pure oxygen what mass is in excess what mass of mgo is formed 4

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