

6 • The Gas Laws

G A S S T O I C H I O M E T R Y W O R K S H E E T

Please answer the following on separate paper using proper units and showing all work. Please note that these problems require a **balanced chemical equation**.

1. Carbon monoxide reacts with oxygen to produce carbon dioxide. If 1.0 L of carbon monoxide reacts with oxygen at STP,
 - a. How many liters of oxygen are required to react?

 - b. How many liters of carbon dioxide are produced?

2. Acetylene gas (C_2H_2) undergoes combustion to produce carbon dioxide and water vapor,
 - a. How many liters of C_2H_2 are required to produce 75.0 L of CO_2 ?

 - b. What volume of H_2O is produced?

 - c. What volume of O_2 is required?

3. If liquid carbon disulfide (CS_2) reacts with 450 mL of oxygen to produce the gases carbon dioxide and sulfur dioxide, what volume of each product is produced?

4. Assume that 5.60 L of hydrogen gas at STP reacts with copper (II) oxide according to the following balanced equation:
$$\text{CuO (s)} + \text{H}_2 \text{ (g)} \rightarrow \text{Cu (s)} + \text{H}_2\text{O (g)}$$
 - a. How many moles of H_2 react?

 - b. How many moles of copper are produced?

 - c. How many grams of copper are produced?

Gas Stoichiometry Problem And Answer

Mike Jess

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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