Gas Steichkonsetry Practice 1.0 - Amorea Mer-

Objective: - Perform stoichiometry calculations of gases including those at STP using molar volume

Directions: Solve such as stochometry problem. Show all work, litelade units

 Calculate the volume of hydrogen gas that well react with exactly 4.6 L of swygen gas using the equation below. Assume there is no change in temperature or pressure.

What volume of CO₂ gas will be produced from U.S.L. of oxygen gas reacting with excess others?

3. What volume of hydrogen at STP can be produced from the reaction of 36.1 g of your with hydrochiosic scul?

 Store many grams of NaCl can be produced by the reaction of 12 d L of chlorine at STP with an excess of section?

 In order to produce 50.0 lines of animous gas (NOL) at NTP, how many grams of both hydrogen and satrogen would you need to react?

Gas Stoichiometry Practice Problems Answer Key

Rachel S Tattersall

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