Gas Stoichiometry #1

If reactants and products are at the same conditions of temperature and pressure, then mole ratios of <u>gases</u> are also volume ratios.

$$3 H_2(g) + N_2(g) \rightarrow 2NH_3(g)$$

- 3 moles H_2 + 1 mole N_2 \rightarrow 2 moles NH_3
- 3 liters H_2 + 1 liter N_2 \rightarrow 2 liters NH_3

Gas Stoichiometry Edgenuity

Sebastian Brünink

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