

Part II: Kinetic Energy and Speed

Sketch and compare the distributions for kinetic energy and speed at two different temperatures in the table below. Record your temperatures (T_1 and T_2), set Volume as a Constant Parameter, and use roughly the same number of particles for each experiment (aim for ~100-200). Use the T_2 temperature to examine a mixture of particles.

Tips:

$T_1 =$ _____ K

The Species Information and Energy Histograms tools will help.

$T_2 =$ _____ K

The system is dynamic so the distributions will fluctuate.

Sketch the average or most common distribution that you see.

	"Heavy" Particles Only	"Light" Particles Only	Heavy + Light Mixture
# of particles (~100-200)			
Kinetic Energy Distribution sketch for T_1			
Speed Distribution sketch for T_1			
Kinetic Energy Distribution sketch for T_2			
Speed Distribution sketch for T_2			

Gas Properties Simulation Activity Answers

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the Petrogav International Oil & Gas Training Center text.

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