

THE GAS LAWS



- Boyle's Law: At constant temperature, the volume of a gas is inversely proportional to its pressure.
- Charles's Law: At constant pressure, the volume of a gas is directly proportional to its temperature in Kelvin.
- Gay-Lussac's Law: At constant volume, the pressure of a gas is directly proportional to its temperature in Kelvin.
- Avogadro's Law: Equal volumes of gases, measured under the same conditions of temperature and pressure, contain the same number of molecules.

Gas Laws Calculations For Matriculants

Julia Schneider



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