

Name: _____ Date: _____ Period: _____

SOLUBILITY WORKSHEET

Use the solubility curves on the attached page to answer the following questions.

- These solubility curves are based on the solubility (in grams) of various salts dissolving in _____ grams of water. In general for solids, as temperature increases, the solubility of the salt _____.
- Most and least soluble at different temperatures:
 - Which salt is **least** soluble at 20°C? _____
 - Which salt is **most** soluble at 10°C? _____
 - For which salt does the solubility change the least? _____
- How many grams of each solute below is needed to saturate 100g of water at 80 °C to prepare the following solutions?
 - KNO₃ _____
 - KCl _____
 - NaCl _____
- How many grams of each solute below are needed to make a **saturated** solution at 50 °C?
 - KClO₃ _____
 - NaNO₃ _____
 - KNO₃ _____
- At what temperature are the following solutes equally soluble in 100g of water?
 - NaNO₃ and KNO₃ _____
 - KNO₃ and NaCl _____
 - NaCl and KCl _____
- The quantity of solute per 100 g of water used in making some solutions is listed below. Is the solution **saturated, unsaturated, or supersaturated**?

Solute	Temperature	
a. 103 g KNO ₃	60 °C	_____
b. 40 g KCl	40 °C	_____

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