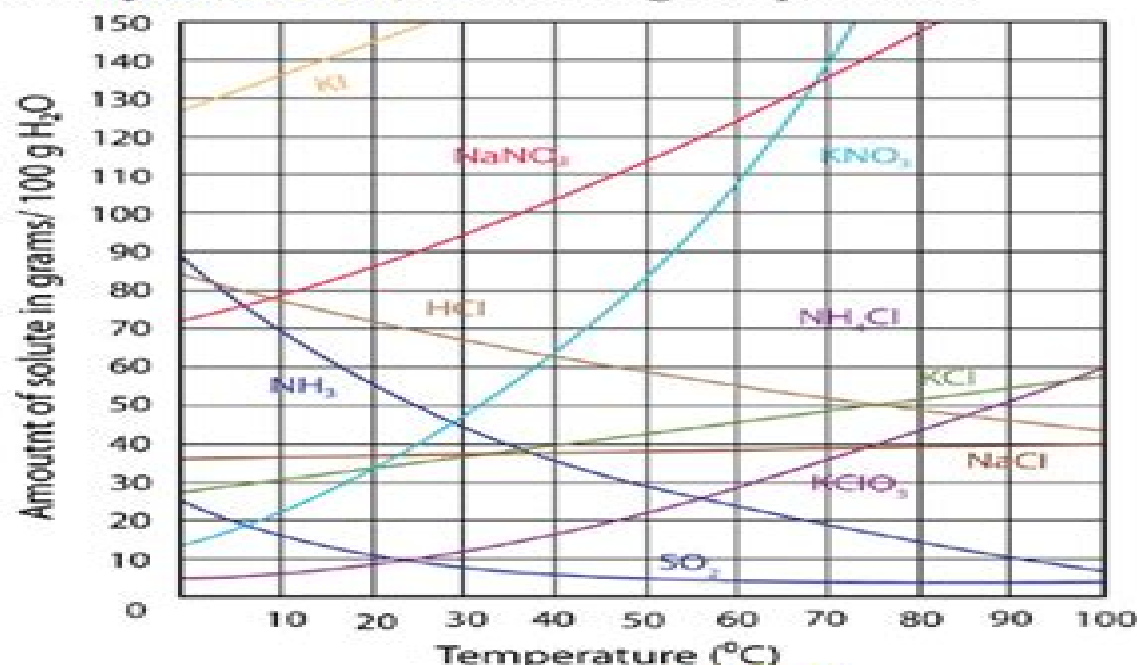




Interpreting Solubility Curve



1. Define solubility. The maximum amount of solute that can dissolve in a solvent
2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve? H₂O
- II. Saturation points are given per how much of the solvent? 10
- III. What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
 10°C = 79 40°C = 103 80°C = 147
- IV. How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
 10°C = 70 30°C = 45 90°C = 10
- V. Which salt is the least soluble in water? NaCl
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C? _____
- VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water? _____
- VIII. Which salt shows the least change in solubility from 0° to 100° C? _____
- IX. Which substance shows a decrease in solubility from 0° to 100° C? _____
- X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? _____

Interpreting Solubility Curves Pogil 2005 Answer Key

N Noddings



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In a global driven by information and connectivity, the power of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such may be the essence of the book **Interpreting Solubility Curves Pogil 2005 Answer Key**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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