

## Chemistry – Mock Test Paper

[Time:  $1\frac{1}{2}$  hrs]

[M. Marks : 80]

Answers to this paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

### Section I (40 Marks)

Attempt all questions from this section

#### Question 1.


(a) Choose the most appropriate answer.

- (i) The element with largest size in second period is :  
 A. Lithium                      B. Oxygen                      C. Fluorine                      D. Carbon
- (ii) Which is the element which belongs to group 13 from :  
 A. Beryllium                      B. Boron                      C. Aluminium                      D. Carbon
- (iii) The most strong alkali is  
 A. Lithium hydroxide                      B. Potassium hydroxide  
 C. Sodium hydroxide                      D. Rubidium hydroxide
- (iv) Lanthanides is a group of elements outside the Periodic Table having atomic number from  
 A. 57 to 71                      B. 89 to 103                      C. 72 to 86                      D. 104 to 118
- (v) The elements with lowest first ionisation energy is  
 A. Sodium                      B. Caesium                      C. Barium                      D. Magnesium
- (vi) When equal volumes of hydrogen gas and chlorine gas are exposed to diffused sunlight the reaction :  
 A. does not take place                      B. takes place at moderate speed  
 C. is explosive in nature                      D. none of the above.
- (vii) In laboratory ammonia gas prepared by heating a mixture of :  
 A.  $\text{NH}_4\text{Cl}$  and  $\text{Ca}(\text{OH})_2$                       B.  $(\text{NH}_4)_2\text{SO}_4$  and  $\text{Ca}(\text{OH})_2$   
 C.  $\text{NH}_4\text{NO}_3$  and  $\text{KOH}$                       D.  $\text{NH}_4\text{NO}_3$  and  $\text{Ca}(\text{OH})_2$
- (viii) The compound which will liberate carbon dioxide on treating with dilute sulphuric acid is :  
 A.  $\text{MgCl}_2$                       B.  $\text{Na}_2\text{SO}_4$                       C.  $\text{Na}_2\text{CO}_3$                       D.  $\text{Na}_2\text{SO}_4$
- (ix) Which is not a member of alkanolic series?  
 A.  $\text{HCOOH}$                       B.  $\text{C}_2\text{H}_5\text{COOH}$                       C.  $\text{HCHO}$                       D.  $\text{CH}_3\text{COOH}$
- (x) Ethyne can be converted into ethyne oxide by treating with :  
 A. Hydrogen peroxide                      B. Ozone  
 C. Oxygen                      D. Conc.  $\text{HNO}_3$

[10]

# Grade 10 Chemistry Paper November Exam

**Boston (Mass.)**



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