QUESTION 5 (Start on a new page.)

The table below shows the melting and boiling points of certain elements. Study table and answer the questions that follow:

| Element | Molting Point (*C) | Boiling Point (°C) |
|---------|--------------------|--------------------|
| Copper | 1 083 | 2 567 |
| Sulphur | 113 | 445 |
| Helium | -272 | -269 |
| Carbon | 3 500 | 4 B27 |

Define to the solid point it will dis.

- 5.2 Identify the change in phase when a liquid changes to become a solid.
- 5.3 In what state of matter will each of the substances be at room temperate
 - 5.3.1 Hellum
 - 5.3.2 Carbon.
- 5.4 Identify and explain the elements with the following types of forces between at atoms:
 - 5.4.1 weakest

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