

Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are not Simple Ions

Number

In substituted compounds, which are formed by adding substituents (atoms and functional groups) to form substances (molecules).

Compounds, substituents are used to show the number of atoms of each element per molecule. This is often done by using the prefix and the number of atoms each represents:

mono	di	tri	tetra	penta	hexa	hepta	octa	nona	deca
1	2	3	4	5	6	7	8	9	10

Write the formula of these substituted compounds:

carbon tetrachloride CCl_4

sulfur dioxide SO_2

dinitrogen monoxide N_2O

dinitrogen trioxide N_2O_3

dinitrogen pentoxide N_2O_5

nitric oxide NO

phosphorus trichloride PCl_3

carbon disulfide CS_2

triphosphorus dioxide P_3O_2

carbon monoxide CO

Organic Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded

Carbon. The number of the carbon atoms each has of

carbon atoms in an organic compound will determine whether the compound is saturated or unsaturated. This is often done by using the prefix and the number of atoms each represents:

monosaturated compounds are simple, linear, and are of monosaturated compounds.

Other kinds of organic compounds are unsaturated.

Simple compounds. The organic compounds, containing only hydrogen and

oxygen are hydrocarbons.

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Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Decoding **Gpb Chemistry Note Answers**: Revealing the Captivating Potential of Verbal Expression

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