Name:

GAS LAWS

Across

- For a gas, the volume that one mole occupies at 0 Celsius and 1 atm pressure.
- 9. The last this techs for a find nost massive rital temperature, the volume of the gas occreases as the pressure of the gas increases nor identical masses of gas, at constant temperature
- 10. Process by which particles under pressure pass through a tiny opening.
- 11. Units of P depend on units of measurement used for P, Y and TR = .0821
- 13. A gas lat or no be live colored by locating to the sulptions of the kinetic-molorated or
- State: hat equal viring figures at the state of particles.
- An imaginary gas that perfectly fits all the assumptions of the kinetic-molecular theory.

Down

- 1. Zero (the kr vin cal which n eser the lowest sso theoretical temperature; atoms are all in the lowest possible energy state.
- 2. An instrument that measures atmospheric pressure
- The pressure of earth's atmosphere at sea level; exactly equivalent to 760 mm
 Hg
- 4. Gas p tir si sil .lo --- t c aa oth r
- The last that site the presse of garage at a cc. an source is directly proportional to the absolute temperature,
- A collision between gas particles and between gas particles and container walls in which there is no net loss of kinetic energy.
- 8. The matter relation is among pass with me temperature, and the number of oir oil ga
- 12. The sane is and acy of a unstar elt move on a oncentration gradient from a mondouve......d to diess conce......ted dies.
- 14. The relationship between pressure, volume, and temperature of a fixed amount of gas.
- As Volume increases temperature increases if pressure is held constant.

Gas Laws Vocabulary Chart Answers

Lewis Morris

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