

Gas laws practice test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. Pressure is the force per unit
a. volume. c. length.
b. surface area. d. depth.
- ____ 2. Why does a can collapse when a vacuum pump removes air from the can?
a. The inside and outside forces balance out and crush the can.
b. The unbalanced outside force from atmospheric pressure crushes the can.
c. The atmosphere exerts pressure on the inside of the can and crushes it.
d. The vacuum pump creates a force that crushes the can.
- ____ 3. If the height of mercury in a barometer at 0°C is less than 760 mm Hg, then
a. the atmospheric pressure is less than standard atmospheric pressure.
b. the atmospheric pressure is greater than standard atmospheric pressure.
c. the atmospheric pressure is equal to standard atmospheric pressure.
d. the atmospheric pressure cannot be determined.
- ____ 4. Convert the pressure 0.840 atm to mm Hg.
a. 365 mm Hg c. 638 mm Hg
b. 437 mm Hg d. 780 mm Hg
- ____ 5. Convert the pressure 1.30 atm to kPa.
a. 2 kPa c. 132 kPa
b. 115 kPa d. 245 kPa
- ____ 6. Standard temperature is exactly
a. 100°C. c. 0°C.
b. 273°C. d. 0 K.
- ____ 7. Three samples of gas each exert 740. mm Hg in separate 2 L containers. What pressure do they exert if they are all placed in a single 2 L container?
a. 247 mm Hg c. 1.48×10^3 mm Hg
b. 740 mm Hg d. 2.22×10^3 mm Hg
- ____ 8. If the temperature remains constant, V and P represent the original volume and pressure, and V' and P' represent the new volume and pressure, what is the mathematical expression for Boyle's law?
a. $P'V = V'P$ c. $VP = V'P'$
b. $VV' = PP'$ d. $V' = \frac{VP}{P'}$
- ____ 9. A sample of oxygen occupies 560. mL when the pressure is 800.00 mm Hg. At constant temperature, what volume does the gas occupy when the pressure decreases to 700.0 mm Hg?
a. 80.0 mL c. 600. mL
b. 490. mL d. 640. mL
- ____ 10. If the temperature of a fixed quantity of gas decreases and the pressure remains unchanged,
a. its volume increases. c. its volume decreases.
b. its volume is unchanged. d. its density decreases.

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Table of Contents Gas Laws Thermodynamics And Kinetics Practice Test

1. Understanding the eBook Gas Laws Thermodynamics And Kinetics Practice Test
 - The Rise of Digital Reading Gas Laws Thermodynamics And Kinetics Practice Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Laws Thermodynamics And Kinetics Practice Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Laws Thermodynamics And Kinetics Practice Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Laws Thermodynamics And Kinetics Practice Test

- Personalized Recommendations
 - Gas Laws Thermodynamics And Kinetics Practice Test User Reviews and Ratings
 - Gas Laws Thermodynamics And Kinetics Practice Test and Bestseller Lists
5. Accessing Gas Laws Thermodynamics And Kinetics Practice Test Free and Paid eBooks
 - Gas Laws Thermodynamics And Kinetics Practice Test Public Domain eBooks
 - Gas Laws Thermodynamics And Kinetics Practice Test eBook Subscription Services
 - Gas Laws Thermodynamics And Kinetics Practice Test Budget-Friendly Options
 6. Navigating Gas Laws Thermodynamics And Kinetics Practice Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Laws Thermodynamics And Kinetics Practice Test Compatibility with Devices
 - Gas Laws Thermodynamics And Kinetics Practice Test Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Laws Thermodynamics And Kinetics Practice Test
 - Highlighting and Note-Taking Gas Laws Thermodynamics And Kinetics Practice Test
 - Interactive Elements Gas Laws Thermodynamics And Kinetics Practice Test
 8. Staying Engaged with Gas Laws Thermodynamics And Kinetics Practice Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Laws Thermodynamics And Kinetics Practice Test
 9. Balancing eBooks and Physical Books Gas Laws Thermodynamics And Kinetics Practice Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Laws Thermodynamics And Kinetics Practice Test
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Gas Laws Thermodynamics And Kinetics Practice Test
 - Setting Reading Goals Gas Laws Thermodynamics And Kinetics Practice Test
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Gas Laws Thermodynamics And Kinetics Practice Test

- Fact-Checking eBook Content of Gas Laws Thermodynamics And Kinetics Practice Test
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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