

Gas laws practice test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. Pressure is the force per unit
a. volume. c. length.
b. surface area. d. depth.
- ____ 2. Why does a can collapse when a vacuum pump removes air from the can?
a. The inside and outside forces balance out and crush the can.
b. The unbalanced outside force from atmospheric pressure crushes the can.
c. The atmosphere exerts pressure on the inside of the can and crushes it.
d. The vacuum pump creates a force that crushes the can.
- ____ 3. If the height of mercury in a barometer at 0°C is less than 760 mm Hg, then
a. the atmospheric pressure is less than standard atmospheric pressure.
b. the atmospheric pressure is greater than standard atmospheric pressure.
c. the atmospheric pressure is equal to standard atmospheric pressure.
d. the atmospheric pressure cannot be determined.
- ____ 4. Convert the pressure 0.840 atm to mm Hg.
a. 365 mm Hg c. 638 mm Hg
b. 437 mm Hg d. 780 mm Hg
- ____ 5. Convert the pressure 1.30 atm to kPa.
a. 2 kPa c. 132 kPa
b. 115 kPa d. 245 kPa
- ____ 6. Standard temperature is exactly
a. 100°C. c. 0°C.
b. 273°C. d. 0 K.
- ____ 7. Three samples of gas each exert 740. mm Hg in separate 2 L containers. What pressure do they exert if they are all placed in a single 2 L container?
a. 247 mm Hg c. 1.48×10^3 mm Hg
b. 740 mm Hg d. 2.22×10^3 mm Hg
- ____ 8. If the temperature remains constant, V and P represent the original volume and pressure, and V' and P' represent the new volume and pressure, what is the mathematical expression for Boyle's law?
a. $P'V = V'P$ c. $VP = V'P'$
b. $VV' = PP'$ d. $V' = \frac{VP}{P'}$
- ____ 9. A sample of oxygen occupies 560. mL when the pressure is 800.00 mm Hg. At constant temperature, what volume does the gas occupy when the pressure decreases to 700.0 mm Hg?
a. 80.0 mL c. 600. mL
b. 490. mL d. 640. mL
- ____ 10. If the temperature of a fixed quantity of gas decreases and the pressure remains unchanged,
a. its volume increases. c. its volume decreases.
b. its volume is unchanged. d. its density decreases.

Gas Law Test Questions

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Gas Law Test Questions:

Adopting the Song of Expression: An Emotional Symphony within **Gas Law Test Questions**

In some sort of eaten by displays and the ceaseless chatter of fast interaction, the melodic beauty and emotional symphony produced by the prepared term usually diminish into the back ground, eclipsed by the constant sound and disturbances that permeate our lives. But, set within the pages of **Gas Law Test Questions** a marvelous literary prize brimming with natural emotions, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, that captivating masterpiece conducts readers on a psychological trip, well unraveling the concealed songs and profound impact resonating within each carefully constructed phrase. Within the depths of the moving assessment, we shall explore the book is central harmonies, analyze their enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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Table of Contents Gas Law Test Questions

1. Understanding the eBook Gas Law Test Questions
 - The Rise of Digital Reading Gas Law Test Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Law Test Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Law Test Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Law Test Questions
 - Personalized Recommendations

- Gas Law Test Questions User Reviews and Ratings
- Gas Law Test Questions and Bestseller Lists
- 5. Accessing Gas Law Test Questions Free and Paid eBooks
 - Gas Law Test Questions Public Domain eBooks
 - Gas Law Test Questions eBook Subscription Services
 - Gas Law Test Questions Budget-Friendly Options
- 6. Navigating Gas Law Test Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Law Test Questions Compatibility with Devices
 - Gas Law Test Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Law Test Questions
 - Highlighting and Note-Taking Gas Law Test Questions
 - Interactive Elements Gas Law Test Questions
- 8. Staying Engaged with Gas Law Test Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Law Test Questions
- 9. Balancing eBooks and Physical Books Gas Law Test Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Law Test Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gas Law Test Questions
 - Setting Reading Goals Gas Law Test Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gas Law Test Questions
 - Fact-Checking eBook Content of Gas Law Test Questions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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