



## Activity B:

### Voltage gradients

#### Get the Gizmo ready:

- Set the **Wavelength** to 300 nm, the **Photon flux** to 10  $\gamma/\text{ms}$ , and the **Voltage** to 0.0 volts.
- Turn on **Show voltage gradient**.



**Introduction:** The electrons that are freed from the surface of the metal have a specific amount of kinetic energy. Faster electrons have greater energies than slower ones. The energy of emitted electrons is measured by setting up an electrical field that opposes their motion. The **voltage** of the field is a measure of its strength.

**Goal:** Use a voltage gradient to measure the energy of emitted electrons.

1. **Observe:** Check that **Potassium** is selected. Click **Flash the light** and observe the emitted electrons. Increase the **Voltage** to 1.5 volts, and click **Flash the light** again.

How does the electrical field affect the motion of the emitted electrons?

The electrons are caused to slow down as they move through the tube because of the electrical field.

2. **Measure:** The energy of an emitted electron is measured in **electron volts** (eV). An electron with an energy of 1 eV can overcome an electrical field of 1 volt. In the Gizmo, increase the voltage until you find the highest voltage that still allows the electrons to reach the light bulb.

What is this value?

1.8eV

This is equal to the energy of the emitted electrons in eV.

3. **Gather data:** With the **Wavelength** set to 300 nm, measure the energy of emitted electrons for potassium, calcium, and uranium. Then measure the same values with wavelengths of 250 nm and 200 nm to complete the table.

| Element   | Energy of emitted electrons (eV) |        |        |
|-----------|----------------------------------|--------|--------|
|           | 300 nm                           | 250 nm | 200 nm |
| Potassium | 1.8eV                            | 2.6eV  | 3.8eV  |
| Calcium   | 1.2eV                            | 2eV    | 3.2eV  |
| Uranium   | 0.5eV                            | 1.3eV  | 2.5eV  |

4. **Analyze:** What patterns do you notice in your data?

The wavelength of the light is decreased as the energy of released electrons increases.

5. **Infer:** Based on your data, which element is hardest to extract electrons from?

Uranium

Explain

The electrons emitted from uranium has the least energy at each wavelength

# Explore Learning Gizmo Photoelectric Effect Answers

**Emilie Sanchez**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Explore Learning Gizmo Photoelectric Effect Answers:**

## **Explore Learning Gizmo Photoelectric Effect Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Explore Learning Gizmo Photoelectric Effect Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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### **Table of Contents Explore Learning Gizmo Photoelectric Effect Answers**

1. Understanding the eBook Explore Learning Gizmo Photoelectric Effect Answers
  - The Rise of Digital Reading Explore Learning Gizmo Photoelectric Effect Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Gizmo Photoelectric Effect Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Explore Learning Gizmo Photoelectric Effect Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Gizmo Photoelectric Effect Answers
  - Personalized Recommendations
  - Explore Learning Gizmo Photoelectric Effect Answers User Reviews and Ratings
  - Explore Learning Gizmo Photoelectric Effect Answers and Bestseller Lists

5. Accessing Explore Learning Gizmo Photoelectric Effect Answers Free and Paid eBooks
  - Explore Learning Gizmo Photoelectric Effect Answers Public Domain eBooks
  - Explore Learning Gizmo Photoelectric Effect Answers eBook Subscription Services
  - Explore Learning Gizmo Photoelectric Effect Answers Budget-Friendly Options
6. Navigating Explore Learning Gizmo Photoelectric Effect Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Explore Learning Gizmo Photoelectric Effect Answers Compatibility with Devices
  - Explore Learning Gizmo Photoelectric Effect Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Explore Learning Gizmo Photoelectric Effect Answers
  - Highlighting and Note-Taking Explore Learning Gizmo Photoelectric Effect Answers
  - Interactive Elements Explore Learning Gizmo Photoelectric Effect Answers
8. Staying Engaged with Explore Learning Gizmo Photoelectric Effect Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Explore Learning Gizmo Photoelectric Effect Answers
9. Balancing eBooks and Physical Books Explore Learning Gizmo Photoelectric Effect Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Explore Learning Gizmo Photoelectric Effect Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Gizmo Photoelectric Effect Answers
  - Setting Reading Goals Explore Learning Gizmo Photoelectric Effect Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Gizmo Photoelectric Effect Answers
  - Fact-Checking eBook Content of Explore Learning Gizmo Photoelectric Effect Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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