

Answers to Practice Exam Questions: Glycolysis

(Hint: see textbook pages 306-307 for a summary of the reactions in glycolysis.)

1. Which of the following phosphorylated compounds would you predict to be a "low-energy" intermediate in metabolism, relative to the others?
 - A. ATP
 - B. 1,3-bisphosphoglycerate
 - C. Glucose 6-phosphate
 - D. Phosphoenolpyruvate
 - E. NADH

2. Which pair below correctly matches an enzyme with its effector?
 - A. Phosphofructokinase-1 / fructose-1,6-bisphosphate
 - B. Pyruvate kinase / phosphoenolpyruvate
 - C. Hexokinase / glucose 1-phosphate
 - D. Pyruvate kinase / fructose-1,6-bisphosphate
 - E. Hexokinase / fructose-1,6-bisphosphate

3. Which of the following does NOT occur in the reactions of glycolysis between fructose-1,6-bisphosphate and 1,3-bisphosphoglycerate?
 - A. Phosphoryl transfer
 - B. An isomerization
 - C. The formation of a high-energy intermediate
 - D. Oxidation
 - E. All of these occur.

4. In the oxidation reaction of glycolysis, which of the following statements is TRUE?
 - A. It is an irreversible reaction.
 - B. It is a regulated step of glycolysis.
 - C. During the catabolism of glucose to pyruvate, NAD⁺ is oxidized.
 - D. Some of the energy from the oxidation reaction is captured by the formation of ATP.
 - E. The reaction halts in the absence of inorganic phosphate.

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