

Use your notes from the Atomic Structure program to answer the following questions.

1. The atomic number tells the number of positively charged protons in the nucleus of an atom. The atom is neutral because this is also the number of negatively charged electrons in the atom.
2. The mass number tells the total number of protons and neutrons in the nucleus of an atom. These particles collectively are called nucleons since both are located in the nucleus.
3. Isotopes are atoms of the same element with different numbers of neutrons which results in different mass numbers.
4. Write the hyphen notation of the three isotopes of hydrogen:
  - a. Hydrogen - 1  ${}^1_1\text{H}$
  - b. Hydrogen - 2  ${}^2_1\text{H}$
  - c. Hydrogen - 3  ${}^3_1\text{H}$
5. Fill in the following chart (Hint: Isotopes may have different masses than the periodic table lists!):

Symbol	Atomic #	Mass #	# of protons	# of electrons	# of neutrons
Hg	80	201	80	80	121
Mo	42	96	42	42	54
Cu	29	64	29	29	35
C	6	14	6	6	8
	15	31	15	15	16
Pb	82	208	82	82	126
Na	11	24	11	11	13

# Glencoe Section 1 Structure Of The Atom Worksheet Answers

**Peipei Pang**



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## **Table of Contents Glencoe Section 1 Structure Of The Atom Worksheet Answers**

1. Understanding the eBook Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - The Rise of Digital Reading Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Personalized Recommendations
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers User Reviews and Ratings
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers and Bestseller Lists
5. Accessing Glencoe Section 1 Structure Of The Atom Worksheet Answers Free and Paid eBooks
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers Public Domain eBooks
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers eBook Subscription Services
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers Budget-Friendly Options

6. Navigating Glencoe Section 1 Structure Of The Atom Worksheet Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers Compatibility with Devices
  - Glencoe Section 1 Structure Of The Atom Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Highlighting and Note-Taking Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Interactive Elements Glencoe Section 1 Structure Of The Atom Worksheet Answers
8. Staying Engaged with Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Glencoe Section 1 Structure Of The Atom Worksheet Answers
9. Balancing eBooks and Physical Books Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Glencoe Section 1 Structure Of The Atom Worksheet Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Setting Reading Goals Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Fact-Checking eBook Content of Glencoe Section 1 Structure Of The Atom Worksheet Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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