

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte, 70% Gesamtpunkte)

Die nachstehenden Probleme betreffen die Veranschaulichung des Verhaltens der Stoffe (Körper) im Diagramm links. Legen Sie die Zustände fest, die die Stoffe während der angegebenen Vorgänge annehmen.

$$p_1 = p_2 = p_3 = p_4 = p_5 = p_6$$

1. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

$$Q = - \frac{1}{2} \cdot \frac{p_1 \cdot V_1}{p_2 \cdot V_2} \cdot \ln \left(\frac{p_1}{p_2} \right) = - 1 \text{ kJ/mol}$$

2. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

$$Q = \frac{1}{2} \cdot \frac{p_1 \cdot V_1}{p_2 \cdot V_2} \cdot \ln \left(\frac{p_1}{p_2} \right) = 1 \text{ kJ/mol}$$

3. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

$$Q = \frac{1}{2} \cdot \frac{p_1 \cdot V_1}{p_2 \cdot V_2} \cdot \ln \left(\frac{p_1}{p_2} \right) = 1 \text{ kJ/mol}$$

4. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

$$Q = \frac{1}{2} \cdot \frac{p_1 \cdot V_1}{p_2 \cdot V_2} \cdot \ln \left(\frac{p_1}{p_2} \right) = 1 \text{ kJ/mol}$$

5. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

$$Q = \frac{1}{2} \cdot \frac{p_1 \cdot V_1}{p_2 \cdot V_2} \cdot \ln \left(\frac{p_1}{p_2} \right) = 1 \text{ kJ/mol}$$

6. Ein Stoff wird isochor (Volumen $V = 1$ l) von 100 °C auf 200 °C erhitzt und bei konstantem Volumen abgekühlt. Wie groß ist die Wärmemenge Q ?

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Arturo Cuomo



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