

Gizmo Assignment: Balancing Chemical Equations

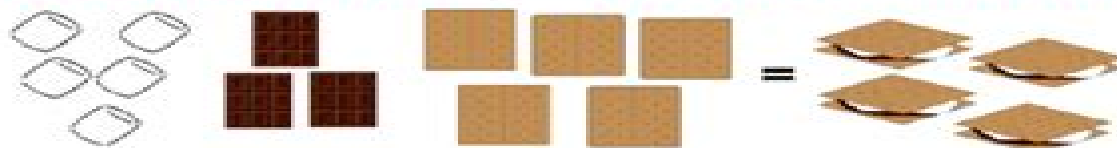
Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The scouts are making s'mores out of toasted marshmallows, chocolate, and graham crackers.

1. What is wrong with the image below?

The materials do not match up with the s'mores in the end. For example, there are 5 marshmallows, 3 chocolate bars, and 5 graham crackers on one side while in the final picture, 8 graham crackers, 4 marshmallows and 4 chocolate bars were used total.



2. Assuming a s'more requires two graham crackers, one marshmallow, and one piece of chocolate, how many s'mores could you make with the ingredients shown? Only 2 total s'mores, with 3 marshmallows, 1 chocolate bar, and 1 graham cracker left over.

Gizmo Warm-up

In a chemical reaction, **reactants** interact to form **products**. This process is summarized by a chemical equation. In the *Balancing Chemical Equations* Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$.

- How many atoms are in a hydrogen **molecule** (H_2)? 2
- How many atoms are in an oxygen molecule (O_2)? 2
- How many hydrogen and oxygen atoms are in a water molecule (H_2O)? 3 total (2 hydrogen + 1 oxygen)
- In general, what does a **subscript** (such as the "2" in H_2) tell you about the molecule?



Gizmo Balancing Chemical Equations Answer Key

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Table of Contents Gizmo Balancing Chemical Equations Answer Key

1. Understanding the eBook Gizmo Balancing Chemical Equations Answer Key
 - The Rise of Digital Reading Gizmo Balancing Chemical Equations Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Balancing Chemical Equations Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Balancing Chemical Equations Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Balancing Chemical Equations Answer Key
 - Personalized Recommendations
 - Gizmo Balancing Chemical Equations Answer Key User Reviews and Ratings

- Gizmo Balancing Chemical Equations Answer Key and Bestseller Lists
- 5. Accessing Gizmo Balancing Chemical Equations Answer Key Free and Paid eBooks
 - Gizmo Balancing Chemical Equations Answer Key Public Domain eBooks
 - Gizmo Balancing Chemical Equations Answer Key eBook Subscription Services
 - Gizmo Balancing Chemical Equations Answer Key Budget-Friendly Options
- 6. Navigating Gizmo Balancing Chemical Equations Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Balancing Chemical Equations Answer Key Compatibility with Devices
 - Gizmo Balancing Chemical Equations Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Balancing Chemical Equations Answer Key
 - Highlighting and Note-Taking Gizmo Balancing Chemical Equations Answer Key
 - Interactive Elements Gizmo Balancing Chemical Equations Answer Key
- 8. Staying Engaged with Gizmo Balancing Chemical Equations Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Balancing Chemical Equations Answer Key
- 9. Balancing eBooks and Physical Books Gizmo Balancing Chemical Equations Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Balancing Chemical Equations Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Balancing Chemical Equations Answer Key
 - Setting Reading Goals Gizmo Balancing Chemical Equations Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Balancing Chemical Equations Answer Key
 - Fact-Checking eBook Content of Gizmo Balancing Chemical Equations Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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