

1. Chemical Foundations

1. A certain volume of a glass of water contains a salt dissolved in a glass of water. Assuming that the glass and the water are completely filled, what happens when the amount of salt is increased and the volume of the glass is decreased? Assume the glass is filled to the brim.

- a. No change
- b. Increased density and mass
- c. No change
- d. Increased density and mass

2. Which of the following is a correct representation of the length of the line representing the density of water?



- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

3. The volume of water is directly proportional to the length of the line representing the density of water. What is the volume of water if the length of the line is 1.000 mL/g?

- a. 1.000 mL
- b. 1.000 g
- c. 1.000 mL/g
- d. 1.000 g/mL

4. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

5. Which of the following is a correct representation of the density of water?



- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

6. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

7. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

8. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

9. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

10. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

11. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

12. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

13. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

14. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

15. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

16. Which of the following is a correct representation of the density of water?

- a. 1.000 g/mL
- b. 1.000 mL/g
- c. 1.000 g/mL
- d. 1.000 mL/g

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