Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3

Section 9.3. Properties of Matter

Label each property as either physical or chemical

- 1. Chemical formula HJO
- 2. Forms green carbonate when exposed to moist air
- is Remains unchanged when in the presence of hitrogen-
- 4. Collord sees
- 5. Solid at normal temperatures and pressures
- 6. Ability to combine with another substance
- 7. Melting point
- Liquid at normal temperatures and pressures.
- a Boiling point is \$60°C
- 56. Conducts electricity
- 11. Density is 1.0 game

For each statement below, write true or Kelse.

- All matter that we encounter in everyday life exists in one of three physical forms.
- 13. A solid has definite shape and volume.
- 24. A liquid has a definite shape and takes on the volume of its container.
- 15. A gas has both the shape and the volume of its container.
- 16. The particles in a gas cannot be compressed into a smaller volume.
- 17. Liquide bend to contract when heated:
- 18. The particles in a solid are spaced for apart.
- 35. The words pay and vapor can be used interchangeably:

Management St. 271

What kinds of changes do these words indicate? Write each word under the correct heading. Use each word only once.

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Exercise 1	The second second	and the second second	-	100	
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200		2.45			
III Consultation of		100 Chillian 1997		DISSOURCE CHESTA	

For each item in Column A. write the letter of the matching item in Column B. Column A.

Columno Bi

- The new substances that are formed in a chemical reaction a chemical change
- 18. A chemical reaction that involves one or more substances. It reactions:

changing into new substances s. s. products

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