



Reinforcement

Structure of the Atom

Directions: Answer the following questions on the lines provided.

1. What is physical and chemical change? (Answered)

Physical change is the change in the physical properties of a substance without changing its chemical composition.

2. What are atoms composed of?

Protons, neutrons, electrons (positive, negative, neutral)

3. How do atoms combine to form molecules? (Answered)

Atoms combine to form molecules by sharing or transferring electrons.

4. What are the types of atoms? (Answered)

1. Metals

5. Why do atoms combine to form molecules? (Answered)

To achieve a stable electronic configuration, atoms combine to form molecules.

6. Why do atoms combine to form molecules? (Answered)

To achieve a stable electronic configuration, atoms combine to form molecules.

7. Why do atoms combine to form molecules? (Answered)

Directions: Answer the questions on the lines provided. (Answered)

Questions 1-10

1.

a. electron

2.

b. neutron

3.

c. proton

4.

d. photon

5.

e. quark

Questions 11-20

11. positively charged particle

12. negatively charged particle

13. neutron

14. particle that carries the positive charge

15. positively charged particle of an atom

Models Of Atom Reinforcement Section Answer

Daniel F McAuley



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