

## Unit 2 Worksheet 1: Molar relationships

### For each of the problems below:

- Write a full balanced chemical equation.
- Calculate, where it applies, molar masses and select one molar volume (moldm<sup>-3</sup> or dm<sup>3</sup>mol<sup>-1</sup>).
- How involved is a more traditional stoichiometric calculation? relatively
- Does it set up for part (c)?

1. Hydrogen sulfide gas, which smells like rotten eggs, burns in air to give gaseous sulfur dioxide and water. How many moles of oxygen gas would be needed to completely burn 5 moles of hydrogen sulfide?

Equation:	$2\text{H}_2\text{S}_{(g)} + \dots \text{O}_{2(g)} \rightarrow \dots \text{SO}_{2(g)} + \dots \text{H}_2\text{O}_{(g)}$
<b>Molar</b>	68                      32                      64                      18
<b>Change</b> 10.0 g    10                      100                      10                      18	

**Answer:**  $2\text{H}_2\text{S}_{(g)} + 3\text{O}_{2(g)} \rightarrow 2\text{SO}_{2(g)} + 2\text{H}_2\text{O}_{(g)}$

2. Propane, C<sub>3</sub>H<sub>8</sub>, burns in air to form carbon dioxide and water. What volume (molar volume was assumed, from an up-to-date of propane data, assumed)?

Equation:	$\text{C}_3\text{H}_{8(g)} + \dots \text{O}_{2(g)} \rightarrow \dots \text{CO}_{2(g)} + \dots \text{H}_2\text{O}_{(g)}$
<b>Molar</b>	44                      32                      44                      18
<b>Change</b> 1.0 g    1.0                      100                      100                      18	

**Answer:**  $\text{C}_3\text{H}_{8(g)} + 5\text{O}_{2(g)} \rightarrow 3\text{CO}_{2(g)} + 4\text{H}_2\text{O}_{(g)}$

3. Oxidation: NO<sub>2</sub> gas is known to react with hydrogen and nitrogen to form an NO<sub>x</sub> hydrocarbon (not a propane). How many moles of a substance can be made from 10.0 moles of nitrogen gas?

Equation:	$\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
Molar	28                      2                      32
Change	10.0 g    10.0                      10.0

# Modeling Chemistry U8 Ws V2 Answers

**Liying Dong**



## **Modeling Chemistry U8 Ws V2 Answers:**

## Decoding **Modeling Chemistry U8 Ws V2 Answers**: Revealing the Captivating Potential of Verbal Expression

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