




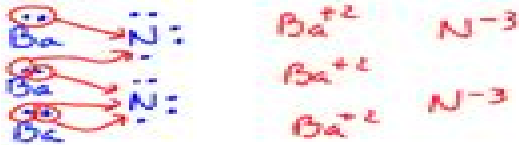








Draw the dot diagrams for each element in the compound

<p>1. NaCl</p>  <p>$\text{Na}^+ \text{Cl}^-$</p>	<p>7. Li_2O</p>  <p>$\text{Li}^+ \text{O}^{2-}$</p>
<p>2. CaO</p>  <p>$\text{Ca}^{+2} \text{O}^{2-}$</p>	<p>8. Ag_3P</p>  <p>$\text{Ag}^+ \text{P}^{3-}$</p>
<p>3. AlN</p>  <p>$\text{Al}^{+3} \text{N}^{3-}$</p>	<p>9. Ba_3N_2</p>  <p>$\text{Ba}^{+2} \text{N}^{3-}$</p>
<p>4. MgS</p>  <p>$\text{Mg}^{+2} \text{S}^{2-}$</p>	<p>10. Al_2O_3</p>  <p>$\text{Al}^{+3} \text{O}^{2-}$</p>
<p>5. FeCl_2</p>  <p>$\text{Fe}^{+2} \text{Cl}^-$</p>	<p>11. ZnI_2</p>  <p>$\text{Zn}^{+2} \text{I}^{-1}$</p>
<p>6. CoF_3</p>  <p>$\text{Co}^{+3} \text{F}^-$</p>	<p>12. Au_2S_3</p>  <p>$\text{Au}^{+3} \text{S}^{2-}$</p>

Modeling Lewis Dot Structures Answer Key

Sabine Zange



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