

- 12 A sample of propanone, CH_3COCH_3 , was heated under reflux with potassium dichromate(VI) acidified with sulfuric acid, and then the mixture was distilled. Apart from the peaks due to the C—C and C—H bonds, what peak(s) would be present in the IR spectrum of the distillate?

- ☐ A A peak due to the C=O only.
☐ B A peak due to the O—H only.
☐ C Peaks due to C=O and O—H .
☐ D Peaks due to C—O , C=O and O—H .

(Total for Question 12 = 1 mark)

- 13 Which of the following describes the appearance of iodine under the stated conditions?

	Solid	Dissolved in aqueous potassium iodide	Dissolved in a liquid hydrocarbon
☐ A	purple	brown	purple
☐ B	brown	blue-black	yellow
☐ C	shiny grey	brown	purple
☐ D	shiny grey	brown	brown

(Total for Question 13 = 1 mark)

- 14 Why do calculations of global warming exclude the effect of water vapour in the atmosphere?

- ☐ A Water vapour is not a greenhouse gas.
☐ B Water vapour is a much less potent greenhouse gas than carbon dioxide.
☐ C The average concentration of water vapour in the atmosphere is fairly constant.
☐ D The concentration of water vapour in the atmosphere is much lower than that of carbon dioxide.

(Total for Question 14 = 1 mark)

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