



PARTICLE THEORY AND STATES OF MATTER

What is the basic structure of matter? Understanding the nature of matter and the arrangement of particles is fundamental. The arrangement and movement of particles explain various properties of the matter.

Particle Theory is the idea that all matter is made up of tiny particles. These particles have three main properties:

- 1. Particles are very small.** They are too small to be seen with the naked eye. They can only be seen using a microscope.
- 2. Particles are constantly moving.** They are always in motion. The speed and direction of their motion is different in different states of matter.
- 3. Particles attract each other.** There are forces of attraction between particles, which vary depending on the type of matter, solids, liquids, and gases.

States of Matter Matter can exist in three primary states: solids, liquids, and gases. Each state has different properties due to the arrangement and movement of its particles.

- Solids:** Particles are closely packed in a regular arrangement. They have strong forces of attraction between them. They vibrate in fixed positions.
- Liquids:** Particles are closely packed but not in a regular arrangement. They have moderate forces of attraction. They can move past each other.
- Gases:** Particles are far apart and move randomly. They have weak forces of attraction. They move very fast in all directions.

Changes of State Matter can change from one state to another. These changes are called changes of state. They are caused by heating or cooling.

- Melting:** Solid to liquid. Heating provides energy to break the strong forces of attraction in solids.
- Freezing:** Liquid to solid. Cooling removes energy, allowing particles to form a regular arrangement.
- Boiling:** Liquid to gas. Heating provides energy to overcome the forces of attraction in liquids.
- Condensation:** Gas to liquid. Cooling removes energy, allowing particles to form a regular arrangement.



11. How does the arrangement of particles change when a solid is heated? What happens to the particles when a solid is heated?

10. How does the arrangement of particles change when a liquid is heated? What happens to the particles when a liquid is heated?

9. What is the change of state when a solid is heated?

8. What happens to the particles when a solid is heated? How does the arrangement of particles change?

7. How does the arrangement of particles change when a gas is heated?

1. What is particle theory?

2. List all the states of matter. Describe each state and its properties.

3. How does the arrangement of particles change when a solid is heated? What happens to the particles when a solid is heated?

4. How does the arrangement of particles change when a liquid is heated? What happens to the particles when a liquid is heated?

5. What happens when a solid is heated? Describe the change in the arrangement of particles.

6. How does the arrangement of particles change when a gas is heated? Describe the change in the arrangement of particles.

Knowledge Matter Reading Answer Key

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