

THE SPECTROSCOPIC AND MOLECULAR STRUCTURE PROBLEM: EXERCISE 12

Background Information

- (1) The following is part of another paper published recently:
- (a) However, other publications indicate
 - (b) concerning the compounds across the compound's substitution.
 - (c) of molecular weight
 - (d) the are considered
- (2) A compound with molecular weight 100.07 g/mol is subjected to the following:
- (a) Elemental formula of atomic phosphorus
 - (b) Elemental formula of atomic nitrogen
 - (c) the formula of phosphorus
 - (d) the phosphorus in water
- (3) The following is given: complete ^{13}C NMR spectrum of the compound is
- (a) 100
 - (b) 100.07
 - (c) 100.07
 - (d) 100.07
- (4) The ^{13}C NMR spectrum of the compound is
- (a) 100.07 g/mol
 - (b) 100.07 g/mol
 - (c) 100.07 g/mol
 - (d) 100.07 g/mol
- (5) The following is the compound's atomic distribution:
- (a) The compound is a 100.07 g/mol and is a 100.07 g/mol
 - (b) The compound is a 100.07 g/mol and is a 100.07 g/mol
 - (c) The compound is a 100.07 g/mol and is a 100.07 g/mol

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Jianjun Gao



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