

Atomic Structure/Isotope Practice Worksheet

1. Fill in the following chart:

Element	# of protons	# of electrons	# of neutrons	Mass number	Atomic number
	16			34	
Rh			72		
Ni					
			56	92	

2. Chlorine has 2 stable isotopes with masses of 35.0 amu (75.53%) and 37.0 amu (24.47%). Calculate the average atomic mass of chlorine.

3. The average atomic mass of Li is 6.941 amu. If the isotopes of lithium have masses of 6.0 and 7.0 amu respectively, what is the relative abundance of each isotope?

4. Calculate the average atomic mass of magnesium.

Isotope	Mass (amu)	Abundance %
^{24}Mg	24	78.99%
^{25}Mg	25	10.00%
^{26}Mg	26	11.01%

Isotopes And Average Atomic Mass Answer Key

**Dr. Subhash Chandra Rastogi, ,Er.
Meera Goyal**



Isotopes And Average Atomic Mass Answer Key:

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EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS

CHANG, 2013-01-07 EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS Chemistry Class 11 - [Bihar & JAC] Dr. S. C. Rastogi, ,Er. Meera Goyal, 2022-07-20 Syllabus Unit I Some Basic Concepts of Chemistry Unit II Structure of Atom Unit III Classification of Elements and Periodicity in Properties Unit IV Chemical Bonding and Molecular Structure Unit V States of Matter Gases and Liquids Unit VI Chemical Thermodynamics Unit VII Equilibrium Unit VIII Redox Reactions Unit IX Hydrogen Unit X s Block Elements Alkali and Alkaline earth metals Group 1 and Group 2 Elements Unit XI Some p Block Elements General Introduction to p Block Elements Unit XII Organic Chemistry Some Basic Principles and Techniques Unit XIII Hydrocarbons Classification of Hydrocarbons Unit XIV Environmental Chemistry Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table Chemistry Class XI - SBPD Publications Dr. Subhash Chandra Rastogi, ,Er. Meera Goyal, 2022-02-17 Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and

Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table *NCERT Chemistry Class 11 - [CBSE Board]* Dr. S. C. Rastogi, ,Er. Meera Goyal,2022-10-11 Syllabus Unit I Some Basic Concepts of Chemistry Unit II Structure of Atom Unit III Classification of Elements and Periodicity in Properties Unit IV Chemical Bonding and Molecular Structure Unit V States of Matter Gases and Liquids Unit VI Chemical Thermodynamics Unit VII Equilibrium Unit VIII Redox Reactions Unit IX Hydrogen Unit X s Block Elements Alkali and Alkaline earth metals Group 1 and Group 2 Elements Unit XI Some p Block Elements General Introduction to p Block Elements Unit XII Organic Chemistry Some Basic Principles and Techniques Unit XIII Hydrocarbons Classification of Hydrocarbons Unit XIV Environmental Chemistry Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I Appendix II Log antilog Table Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series,Albert Tarendash,2021-01-05 Barron s Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day *A-level Chemistry Challenging Learn-By-Example (Yellowreef)* Thomas Bond,Chris Hughes,2015-02-27 according to syllabus for exam up to year 2016 updated new questions from top schools from 2003 end 2013 teachers comments revealing common mistakes wrong habits arrange from easy to hard to facilitate easy absorption most efficient method of learning hence saves time advanced trade book complete edition and concise edition eBooks available also suitable for Cambridge GCE AL H1 H2 Cambridge International AL Cambridge Pre University Books available for other subjects including Physics Chemistry Biology Mathematics Economics English Primary level Secondary level GCE O level GCE A level iGCSE Cambridge A level Hong Kong DSE visit www.yellowreef.com for sample chapters and more

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Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

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