

Question 2**[5]**

You are provided with two organic compounds, C-16 and C-17.

Perform the experiments given below on each of the two compounds. Record the changes taking place at every step of the experiment.

Note the smell of the substance formed, if significant, the colour of the solution obtained, the colour of the precipitate produced and any other observations you may have. State the identity of each compound on the basis of the experiments and observational changes.

PROCEDURE:**(a) Substance C – 16**

- (i) Take 1 ml of C-16 in a test tube and add 1 ml of Fehling's solution. Warm the contents.
- (ii) Take 2 ml of C-16 in a test tube and add 1 ml of lead acetate solution followed by 1 ml of ammonium hydroxide solution. Boil the contents.
- (iii) Take 2 ml of C-16 in a test tube and add 2 to 3 drops of alcoholic α -naphthol solution. Pour conc. sulphuric acid slowly along the side of the test-tube.

Show the results as required to the Visiting Examiner.

(b) Substance C-17

- (i) Take 1 ml of C-17 in a test-tube and add a few drops of water. Now add about 1 ml of sodium hypochlorite solution; Shake the contents.
- (ii) Take 1 ml of C-17 in a test-tube and add a few drops of dilute sulphuric acid. Now add 1 ml of potassium dichromate solution. Shake and warm the contents.
- (iii) Take 1 ml of C-17 in a test-tube and add 1 ml of conc. hydrochloric acid to it. Now, add a few drops of neutral ferric chloride solution and dilute the contents with water.
- (iv) Take 2 ml of C-17 and dilute it with water. To this, add a few drops of bleaching powder solution and shake. Now, add a few drops of ammonium sulphide solution and shake.

Show the results as required to the Visiting Examiner.

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Table of Contents Isc Question Papers 2014

1. Understanding the eBook Isc Question Papers 2014
 - The Rise of Digital Reading Isc Question Papers 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Isc Question Papers 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Isc Question Papers 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Isc Question Papers 2014
 - Personalized Recommendations
 - Isc Question Papers 2014 User Reviews and Ratings
 - Isc Question Papers 2014 and Bestseller Lists
5. Accessing Isc Question Papers 2014 Free and Paid eBooks
 - Isc Question Papers 2014 Public Domain eBooks
 - Isc Question Papers 2014 eBook Subscription Services
 - Isc Question Papers 2014 Budget-Friendly Options

6. Navigating Isc Question Papers 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Isc Question Papers 2014 Compatibility with Devices
 - Isc Question Papers 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Isc Question Papers 2014
 - Highlighting and Note-Taking Isc Question Papers 2014
 - Interactive Elements Isc Question Papers 2014
8. Staying Engaged with Isc Question Papers 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Isc Question Papers 2014
9. Balancing eBooks and Physical Books Isc Question Papers 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Isc Question Papers 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Isc Question Papers 2014
 - Setting Reading Goals Isc Question Papers 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Isc Question Papers 2014
 - Fact-Checking eBook Content of Isc Question Papers 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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