

C4: Chemical changes

- What are the products when metal carbonates react with acids?

Metal carbonate + acid $\rightarrow$ salt + water + carbon dioxide

- What kind of salts are produced by hydrochloric acid?

Chlorides

- What kind of salts are produced by sulphuric acid?

Sulphide / sulphate

- What kind of salts are produced by nitric acid?

Nitride / nitrate

- How can soluble salts be produced?

React a metal / metal oxide / metal hydroxide / metal carbonate with an acid; filter off excess to produce the salt solution

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L Reisser



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aldehydes and ketones draw the structures for the following compounds ethanal

butanone also known as butan 2 one but the 2 is unnecessary and is often left out 2 methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another branched alkane is

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then look at these questions again ii

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4 dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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