$^{\circ}C = (^{\circ}F - 32) *5/9$

 $^{\circ}F = ^{\circ}C * 9/5 + 32$

 $K = ^{\circ}C + 273$

Calculate the Gibbs free energy change (ΔG) for the following chemical reaction:

ATP ADP + Pi

The reaction occurs at 68 °F, the change in heat $(\Delta H) = 19,070$ cal, and the change in entropy $(\Delta S) = 90$ cal/K.

68 °F = 293 K, therefore the equation is set up in the following way:

 $\Delta G = \Delta H - T\Delta S$

 $\Delta G = 19,070 \text{ cal} - (293 \text{ K}) (90 \text{ cal/K})$

 $\Delta G = -7300 \text{ cal} = -7.3 \text{ kcal}$

Calculate the Gibbs free energy change (ΔG) for the following chemical reaction:

glutamate + NH₃ glutamine + H₂O

The reaction occurs at 68 °F, the change in heat (ΔH) = 4103 cal, and the change in entropy (ΔS) = 2.4 cal/K.

66 °F = 293 K, therefore the equation is set up in the following way: $\Delta G = \Delta H - T\Delta S$ $\Delta G = 4103$ cal -293 K (2.4 cal/ K) $\Delta G = 3399.8$ cal =3.4 kcal

3. Would either of the reactions above occur spontaneously? If so, which one(s) and why?

The reaction with ATP in #1. The change in Gibbs free energy (ΔG) is negative for exothermic reactions and can only be negative for spontaneous reactions. However, the activation energy required would still cause this reaction to occur slowly.

4. Are either of the above reactions endergonic? If so, which one(s) and why?

The reaction with glutamate and NH_2 in #2. The change in ΔG (free energy) is positive, meaning free energy is absorbed, making this reaction endergonic.

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields <u>U Can: Physics I For Dummies</u> Steven Holzner, Daniel Funch Wohns, 2015-08-10 Whether you re a student who just needs to know the vital concepts of physics or you re looking for a basic reference tool this is a must have guide Free of ramp up and ancillary material it contains content focused on key topics only provides discrete explanations of critical concepts taught in an introductory physics course and provides a perfect reference for parents who need to review critical physics concepts as they help high school students with homework assignments Chapter-wise NCERT + Exemplar + Practice Questions with Solutions for CBSE Physics Class 11 2nd edition Disha Experts, 2017-08-29 The book Chapter wise NCERT Exemplar Practice Questions with Solutions for CBSE Class 11 Physics has been divided into 3 parts Part A provides detailed solutions Question by Question of all the questions exercises provided in the NCERT Textbook Part B provides solutions to the questions in the NCERT Exemplar book Part C provides selected Practice Questions useful for the Class 11 examination along with detailed solutions. The solutions have been designed in such a manner Step by Step that it would bring 100% Concept Clarity for the student Work, Energy and Power Sanjay Kumar, 2020-09-16 This text book is primarily intended for students who are preparing for the entrance tests of IIT IEE NEET AIIMS and other esteemed colleges in same fields This text is equally useful to the students preparing for their school exams Main Features of the Book 1 Every concept is given in student friendly language with various solved problems and checkpoint questions The solution is provided with problem solving approach and discussion 2 Special attention is given to tricky topics like work energy theorem conservative and non conservative forces conservation of mechanical energy work done by non conservative forces power of pump and chain related problems so that student can easily solve them with fun 3 To test the understanding

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