








Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their relatively different mass. However, when the elements react, combinations formed are consistent in their composition. Why do the elements react, and how are the combinations formed? We will now examine the patterns that exist for the ratios in which these elements combine in order to describe the changes of the basic building blocks.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic symbols and formulas	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$	1 hydrogen 1 oxygen $\frac{1}{1}$
particle diagram				

Atomic symbols and formulas	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$
particle diagram			

Atomic symbols and formulas	12 hydrogen 16 oxygen $\frac{12}{16}$	12 hydrogen 16 oxygen $\frac{12}{16}$	1 hydrogen 16 oxygen $\frac{1}{16}$	1 hydrogen 16 oxygen $\frac{1}{16}$
particle diagram				

Modeling Chemistry Unit 6 Test Answer Key

Siddappa N.Byrareddy



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