

Chemistry Unit 8 Worksheet 4

Examples of Every Kind of Problem

This is a separate sheet of paper, so this is a complete solution to each of the problems. You can follow the problems without an answer. We want to make your final solution.

12. Calculate the number of moles of potassium chlorate, KClO_3 , that must decompose to produce potassium chloride KCl and a total of 1.5 moles of oxygen gas.



Subst:	1.0	0	0
Change:	-1.0	+1.0	+1.5
Final:	0	1.0	1.5

1.0 moles KClO_3 needed

13. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 5.0 g of magnesium?



Subst:	0.200	0	0	0
Change:	-0.200	-0.400	+0.200	+0.200
Final:	0	0	0.200	0.200

0.400 moles HCl needed

14. Ethane, C_2H_6 , reacts with oxygen gas to produce carbon dioxide gas and water vapor. What number of moles of oxygen are required to react with 0.50 moles of ethane?



Subst:	0.50	0.0	0	0
Change:	-0.50	-1.75	+1.00	+1.50
Final:	0	0.0	1.00	1.50

1.75 moles O_2 needed

15. Sodium metal reacts with sulfur dioxide to produce sodium sulfide and sodium sulfate. How many moles of sodium metal are needed to produce 1.00 moles of sodium sulfate?



Subst:	0.500	0.0	0	0
Change:	-0.500	-0.250	+0.500	+1.000
Final:	0	0.0	0.500	1.000

0.500 moles Na needed

Modeling Chemistry Unit 11 Ws1

LL Leslie



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