

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that we are a "paper" writer, a chemist who has to write out what products may be made. We have the reactants and know the mass. The mass is underlined (underlined means).

1. Chlorine gas reacts with sodium metal to produce sodium chloride. NaCl and NaClO_2 are the only products. The products are sodium chloride, NaCl and NaClO_2 (sodium chloride).

Reactants	100.0 g	100.0 g	100.0 g
Products	100.0 g	100.0 g	100.0 g
Mass	100.0 g	100.0 g	100.0 g

100.0 g
100.0 g

2. Chlorine gas reacts with sodium metal to produce sodium chloride. NaCl and NaClO_2 are the only products. The products are sodium chloride, NaCl and NaClO_2 (sodium chloride).

Reactants	100.0 g	100.0 g	100.0 g
Products	100.0 g	100.0 g	100.0 g
Mass	100.0 g	100.0 g	100.0 g

100.0 g
100.0 g

3. Chlorine gas reacts with sodium metal to produce sodium chloride. NaCl and NaClO_2 are the only products. The products are sodium chloride, NaCl and NaClO_2 (sodium chloride).

Reactants	100.0 g	100.0 g	100.0 g
Products	100.0 g	100.0 g	100.0 g
Mass	100.0 g	100.0 g	100.0 g

100.0 g
100.0 g

4. Chlorine gas reacts with sodium metal to produce sodium chloride. NaCl and NaClO_2 are the only products. The products are sodium chloride, NaCl and NaClO_2 (sodium chloride).



100.0 g
100.0 g

Reactants	100.0 g	100.0 g	100.0 g
Products	100.0 g	100.0 g	100.0 g
Mass	100.0 g	100.0 g	100.0 g

100.0 g
100.0 g

Modeling Chemistry Unit 8 Formula Practice

ML Yell



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