

Name: Key
 Date: _____ Block: _____

Stoichiometry and Percent Yield Practice

Answer the following questions about theoretical yield and percent yield. Make sure you show all work, units, and round for sig figs.

1. Use the following equation to answer the questions below.



If I start this reaction with 57.0 grams of nitrogen with excess hydrogen, how many grams of ammonia (NH_3) can theoretically be produced?

$$57.0 \text{ g N}_2 \times \frac{1 \text{ mol N}_2}{28.014 \text{ g N}_2} \times \frac{2 \text{ mol NH}_3}{1 \text{ mol N}_2} \times \frac{17.031 \text{ g NH}_3}{1 \text{ mol NH}_3} = \boxed{69.3 \text{ g NH}_3}$$

If 60.0 grams of ammonia is actually produced, what was my percent yield of ammonia?

$$\frac{60.0}{69.3} \times 100 = \boxed{86.6\%}$$

2. Use the following equation to answer the questions below.



If 11.3 grams of carbon dioxide gas is used in this reaction, how many grams of lithium carbonate can theoretically be produced?

$$11.3 \text{ g CO}_2 \times \frac{1 \text{ mol CO}_2}{44.009 \text{ g CO}_2} \times \frac{1 \text{ mol Li}_2\text{CO}_3}{1 \text{ mol CO}_2} \times \frac{73.87 \text{ g Li}_2\text{CO}_3}{1 \text{ mol Li}_2\text{CO}_3} = \boxed{19.0 \text{ g Li}_2\text{CO}_3}$$

If 14.4 grams of lithium carbonate are actually produced, what was my percent yield?

$$\frac{14.4}{19.0} \times 100 = \boxed{75.8\%}$$

Modeling Chemistry Stoichiometry 2 Percent Yield Answers

John H. Meiser



Modeling Chemistry Stoichiometry 2 Percent Yield Answers:

Selected Solutions for Chemistry, Concepts and Models by Robinson, Odom, and Holtzclaw John H. Meiser, 1992

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Brower,Ronald W. McClard,Ronald C. McClard,1990 **Holt Chemistry** Ralph Thomas Myers,2004 General Chemistry for Engineers Jeffrey Gaffney,Nancy Marley,2017-11-13 General Chemistry for Engineers explores the key areas of chemistry needed for engineers This book develops material from the basics to more advanced areas in a systematic fashion As the material is presented case studies relevant to engineering are included that demonstrate the strong link between chemistry and the various areas of engineering Serves as a unique chemistry reference source for professional engineers Provides the chemistry principles required by various engineering disciplines Begins with an atoms first approach building from the simple to the more complex chemical concepts Includes engineering case studies connecting chemical principles to solving actual engineering problems Links chemistry to contemporary issues related to the interface between chemistry and engineering practices **College Chemistry** Henry Fuller Holtzclaw,William R. Robinson,William Harrison Nebergall,1984

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synthesis and classical fermentations into modern bio catalysis and biosynthesis based on insights in metabolic pathways and enzyme actions **Advances in Bioprocess Engineering** Enrique Galindo, Octavio R. Ramírez, 2013-06-29 Bioprocess engineering plays a key role in the development and optimization of bioprocesses leading to the products of biotechnology A survey of the state of the art in this field is greatly needed This work covers all the essential sub areas and as such is required reading for scientists active in all the disciplines involved in bioprocess engineering This review of basic and applied approaches is brought together by a broad international group of expert authors The work is a reflection of the First International Symposium on Bioprocess Engineering June 1994 However it must be emphasized that the book cannot be perceived as a regular symposium proceedings volume a strict peer review process assures the readers of a high level of quality more than a quarter of the work consists of invited contributions while less than half of the spontaneously submitted manuscripts were accepted for publication Advances in Bioprocess Engineering belongs among the indispensable set of instruments of today s researcher in this field *Publications* United States. National Bureau of Standards, 1971 *NBS Special Publication* ,1968 **Energy Research Abstracts** ,1990 *Index Medicus* ,2003 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings **Catalog of National Bureau of Standards Publications, 1966-1976: pt. 1-2.** **Key word index** United States. National Bureau of Standards, 1978 **Nuclear Science Abstracts** ,1975

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